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DESCRIPTORS: \*Chlorides, Ions, Chemical analysis,  
\*Spectroscopy, \*Colorimetric analysis, Calibration,  
\*Iron compounds, \*Perchlorates, \*Mercury compounds,  
\*Thiocyanates, Nitrates.

An iron perchlorate solution and a mercuric thiocya-  
nate solution were added to a specimen solution and the  
chloride ion was determined by measuring the absorb-  
ency of an orange-colored aqueous solution at  
(Chemistry--Analytical, TT, v. 10, no. 12) (over)

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460 m(mu). It was possible to determine from 0.1 to 15 ppm. of chloride ion speedily and accurately.  
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DESCRIPTORS: \*Chlorides, \*Bromides, Oxidation,  
Spectrophotometers, Quantitative analysis.

A method for the determination of traces of chloride  
in large amounts of bromide was studied. An aqueous  
solution containing a fixed quantity of bromide ion was  
oxidized to potassium permanganate at the time when  
it was heated in nitric acidity. Conditions for the  
quantitative oxidation of bromide ion to bromine were  
sought. After removing bromine by heating the chloride  
(Chemistry--Analytical, TT, v. 10, no. 11) (over)

63-18715

I. Tomonari, A.

Office of Technical Services



ion in the aqueous solution was colorimetrically determined in accordance with the mercuric thiocyanate method. It was possible to speedily and accurately determine from 0.005 to 0.5 weight per cent of chloride ion in bromide ion and from 0.003 to 0.3 per cent of chloride ion in potassium bromide in accordance with this method. (Author)

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